

First-Principles Calculation on CMP Process of Glass Surface by CeO₂ Particle and Design of Alternative Abrasive Grain

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1. Introduction

CeO₂ abrasive grains are employed for high-precision smoothing of glass in various electric appliances. Since Ce yields in only a few regions, development of alternative materials of CeO₂ abrasive grains is necessary. In this study, to clarify chemical mechanical polishing (CMP) mechanism of a glass surface by a CeO₂ abrasive grain, we investigated the polishing processes of the SiO₂ surface by the CeO₂ particle via first-principles calculation. Furthermore, based on our revealed CMP mechanism, we propose the design principles for the alternative abrasive grain of CeO₂ and theoretically design a new abrasive grain.

2. Method

We employed the first-principles calculation package, "DMol³" to calculate the electronic states and structure of the abrasive grain and SiO₂ surface.

3. Results and discussion

To clarify the chemical reaction of the CeO₂ particle with the SiO₂ surface under water environment in the CMP process, we initially calculated the Ce atom charges of the CeO₂ particle and Si-O bond length of the SiO₂ surface by the first-principles calculation, when the particle was pushed down to the surface (Fig. 1). Here, we have revealed that Ce atoms of the commercial CeO₂ abrasive grains are exposed on the surface and take a valence value of +III. Then, we adopted the Ce₂O₃ cluster model since this model expresses the surface state of the abrasive grain. The calculation results show that the charges of the Ce atoms increases by pushing down the Ce₂O₃ cluster to the SiO₂ surface. In addition, the two Si-O bond lengths of the SiO₂ surface increase by the contact with the Ce₂O₃ cluster (Fig. 1(b)). These indicate that the Ce³⁺ donates electrons to the Si-O bonds and it leads to the weakening of the Si-O bonds. Next, we investigated the structural change of the SiO₂ surface which the Ce₂O₃ cluster contacts by approach of a H₂O molecule via the first-principles calculation. After the H₂O molecule dissociates to H⁺ and OH⁻, the Si-O bond is stretched due to the binding with H⁺ and OH⁻. We found out that the Si-O bond is broken and the SiO₂ surface is softened due to the chemical reaction by the H₂O molecule. The mechanical polishing of the softened glass surface occurs later.

Then, we suggest that these chemical reactions by the Ce³⁺ atoms and H₂O molecule enhance the

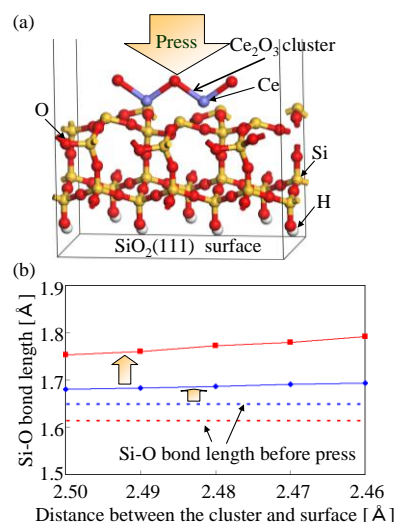


Fig.1 (a) SiO₂(111) surface model pressed by Ce₂O₃ cluster. (b) Change in the Si-O bond length of the SiO₂(111) surface during the press process by Ce₂O₃ cluster.

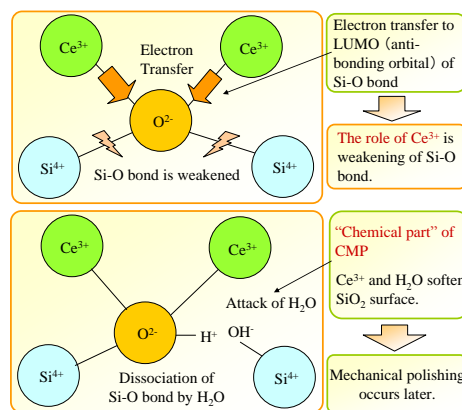


Fig.2 Chemical mechanical polishing mechanism of SiO₂ by CeO₂ abrasive grains.

mechanical polishing processes on the glass surface in the CMP process (Fig. 2). Thus, we can propose that low-valent metal atoms exposed on the surface have chemical reactivity and are efficient for the polishing of the glass surface as a design principle for the alternative materials of CeO₂ abrasive grain.

Based on our established design principle, we proposed CaCeO₃ with the perovskite structure as the alternative materials, in which the more low-valent Ce atoms are exposed on the surface than the CeO₂ abrasive grain. The details of the CMP property of CaCeO₃ will be explained in the conference.